

Veer Narmad South Gujarat University, Surat
B.Sc.

Chemistry Practical
C.B.C.S. (Semester-III)

60 Marks (External)
30 Marks (Internal)

Uni. Exam. 2 days

Gravimetric Estimation of (Any two)

- (1) Fe⁺² as Fe₂O₃ (Given Solution of Fe- NH₄-SO₄ + Conc. H₂SO₄)
(2) Ba⁺² as BaSO₄ (Given Solution of BaCl₂)
(3) Ni⁺² as Ni (DMG)₂ (Given Solution of NiCl₂ + Conc. HCl)

VOLUMETRIC EXERCISE (Any three)

- (1) To determine the amount of Nickel by EDTA.
(2) To determine the amount of Copper by EDTA.
(3) To determine the amount of Zinc by EDTA. .
(4) Determination of total hardness of water by EDTA.

ORGANIC SPOTTING

ACID - Salicylic acid, Cinnamic acid, Phenyl acetic acid, Sulphanilic acid,

BASE - o-Nitroaniline, m-Nitroaniline, p-Nitroaniline, p-Chloroaniline,
Dimethylaniline, Diethylaniline, Diphenylamine, p- toludine.

PHENOL - Alpha naphthol, Beta naphthol, o-Nitro phenol

NEUTRAL- ALDEHYDE - Glucose, Benzaldehyde

KETONE - Methyl Ethyl ketone, Acetophenone

ESTER - Ethylacetate, Butylacetate

ALCOHOL - Ethanol, Butanol

HYDROCARBON - Anthracene, Panhthalene, Diphenyl

NITRO HYDROCARBON - m-Dinitro benzene, Nitro benzene

HALOGENATED HYDROCARBON - Chlorobenzene, Bromobenzene, p-
Dichlorobenzene

AMIDE - Benzamide, Thiourea

ANILIDE - Acetanilide

QUINONE - Anthraquinone

**N.B. Candidate should perform the analysis of at least
8 substances.**

PHYSICAL PRACTICALS : (Any three)

1 Adsorption :

To study the adsorption of given organic acid (acetic acid/oxalic acid) on animal charcoal.

2 Chemical Kinetics - Ester hydrolysis :

To study the monomolecular reaction in the hydrolysis of methyl acetate in 0.5 N HCl at different initial concentrations.

3 Saponification :

To investigate the reaction in saponification of Ethyl acetate by NaOH.

4 . Conductometric Titration

To determine the normality of the given HCl solution by the conductometric titration with the given 0.1 N NaOH solution.

5 Viscosity :

To determine the viscosity of the given liquids and the % of unknown mixture 'C'.

